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Bulky redox-active metalloligands stabilize the phosphorus radical $[\cdot\text{P}\{\text{N}=\text{V}[\text{N}(\text{Np})\text{Ar}]_3\}_2]$ (Np=neopentyl) to the extent that it exists as a monomer even in the solid state. C. C. Cummins and co-workers describe in their Communication on page 3111 ff. the EPR spectrum of the radical (see picture; P orange, N blue, V yellow), which reveals delocalization of the unpaired electron onto both vanadium centers. Radical bond formation at the phosphorus atom leads to diamagnetic products.

